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stone 16 5/1/03
6TPMS
20 ZPO - 650 MM
filter

15 thioxanthone
7 sp ep
9 el. 100ml VM
11 1st color
Crystal violet
Cryspal

(54) Title: RADIATION CURABLE RESIN COMPOSITION FOR MAKING COLORED THREE DIMENSIONAL OBJECTS

(57) Abstract: A radiation curable resin composition suitable for making three dimensional objects comprising at least one epoxy compound, a cationic photoinitiator, wherein the resin composition has a first color or no color before cure and wherein a three dimensional object made from the resin by subjecting the resin to radiation shows a second color which is different from the color of the resin composition before cure.

RADIATION CURABLE RESIN COMPOSITION FOR MAKING COLORED THREE
5 DIMENSIONAL OBJECTS

BACKGROUND OF THE INVENTION

Field of the invention

10 The present invention relates to resin compositions suitable for
making colored three dimensional objects and to methods for making colored three
dimensional objects.

Description of related art

15 Methods for producing three dimensional objects are known in the
field. An example of such a method is stereolithography. This method comprises in
general the steps of irradiating a liquid surface of a liquid photo-curable resin
composition kept in a container with ultraviolet laser beams selectively under the
control of a computer so as to obtain a desired pattern to cure it to a predetermined
thickness, supplying the liquid photo-curable resin composition in an amount enough to
20 form one layer thereof on top of the cured layer, irradiating it with ultraviolet laser
beams likewise to cure the resin so as to form a continuous and integral cured layer,
and repeating this lamination procedure until a three-dimensional object having a final
shape is obtained. In this case, even if the shape of the model is rather complex, a
desired three-dimensional object can be produced easily in a relatively short period of
25 time. Therefore, this method has been attracting much attention in recent years. Three
dimensional objects may also be obtained when the resin is present in the form of a
paste or high viscosity liquid. In that case alternative methods of application of the
successive layers may be applied. Also other ways of curing of successive layers may
be applied, like for example thermal curing or curing with visible light.

30 In recent years, three-dimensional objects obtained by the optical
stereolithography have been developed from concept models to practical applications
such as test models, prototypes or the like. Along with this development, the three-
dimensional objects have been required to have high dimensional accuracy, excellent
shape stability and excellent mechanical properties. In addition, along with
35 development of application and expansion of demand for three-dimensional objects by
optical stereolithography, colored three-dimensional objects, which give a massive
impression, which are colorful and also have the above excellent properties or,

depending on an application, colored three-dimensional objects having light-shielding properties are now in great demand.

Object of the invention

5 The present invention provides radiation curable compositions suitable for use in a process to make colored three dimensional objects by exposing the resin to radiation. Furthermore the present invention provides a process in which relatively low amounts of irradiation may be used for effecting both cure of the resin and coloring of the object.

10

Summary of the invention.

In one embodiment, the present invention provides a radiation curable resin composition suitable for making three dimensional objects comprising at A) least one epoxy compound and B) a cationic photoinitiator, wherein the resin

15 composition has a first color or no color before cure and wherein a three dimensional object made from the resin by subjecting the resin to radiation shows a second color which is different from the color of the resin.

20 A second embodiment of the present invention provides an opaque radiation curable composition suitable for making three dimensional objects comprising a radiation curable component, a photoinitiator and a filler, wherein the resin composition has a first color or no color before cure and wherein a three dimensional object made from the resin by subjecting the resin to radiation shows a second color which is different from the color of the resin.

25 The resin composition of the present invention may show the natural color of a typical resin that is known in the field, ranging from yellow or amber to completely colorless for transparent resins or white for filled resins. Preferably the resin is substantially colorless, which means that the resin composition shows an absorbance of visible light in the region between 400 and 650 nm of less than 0.3, preferably less than 0.2, more preferably less than 0.1, measured on a sample having 30 a thickness of 1 cm. The absorbance is measured on a UV-VIS spectrophotometer in accordance with ASTM E1164-94.

35 In the case that the resin contains a filler, measurement of color with light transmittance is impossible, since the filler will reflect the light. In that case measuring reflectance of light will give good results. In case that the resin composition is for example transparent, but the three dimensional object is opaque, than different

methods may be used to measure either absorbance (for a clear resin composition or three dimensional object) or reflectance (for an opaque object or resin composition, or for example for a resin or object that contains a filler). These methods and principles are all described in ASTM E 1164-94. The reflectance measurement is carried out with 5 a chroma meter, or color spectrophotometer.

The resin composition of the present invention preferably contains a latent coloring component C) that is a substantially colorless dye precursor capable of forming a chromophore in the presence of reactive components in the resin that are formed during irradiation of the resin. It is believed that especially the photoacid that is 10 formed during irradiation of the cationic photoinitiator is one of the active components that reacts with the latent coloring component C) to generate a chromophore that gives color to the resin composition or to the three dimensional part.

The presence of the latent coloring component C) has at least two advantages. One advantage is that the resin composition has a first color (or no color), 15 while the part shows a distinctly different second color. This difference in color facilitates the observation of the part building process, is appealing to the eye and gives a part that is colored throughout the whole part. This will reduce the time that is used in finishing the part, for example by removing support structures used to build the part and for example by surface coloring or painting of the product for a customer. 20 Another advantage is also that it is easier to clean off excess resin because the part and resin are easily differentiated. Thus the cleaning of the part can be more complete and effective. This is especially advantageous when filled resin compositions are used that have high viscosities or are paste-like.

Another advantage is that the quality of the resin composition can be 25 monitored by measuring the color of the resin in the vat of the SL-machine. In the ideal situation, the appearance of the resin will be colorless or the resin will have its natural yellowish or amber appearance for transparent resins, or white for filled resins. Due to stray and transmitted radiation from normal part building some strong acid (cationic initiating species) may be formed in the vat. This strong acid can start, albeit at slow 30 rate, chain addition polymerization of epoxide moieties in the resin, and because this polymerization does not have a naturally occurring chain termination process (as opposed to free radical chain addition polymerization, where bimolecular radical reactions such as coupling or disproportionation can terminate the chain process) the slow chain addition polymerization can continue and ultimately increases resin viscosity 35 to a point where part preparation and clean-up is virtually impossible and the resin

must be replaced. It is believed that this photoacid will react with the component C) in the resin upon formation of a colored complex. So the resin will start showing a slight color when the amount of photoacid increases. The color change is an indicator of a viscosity increase of the resin composition. The color change can be easily reversed by 5 adding a neutralizing amount of a hydroxylic base, e.g., tetramethyl ammonium hydroxide in an organic solvent. This base will irreversibly react with the photoacid and thereby stop the polymerization reaction and viscosity build of the resin. Also the color of the resin, which is believed to be due to the (reversible) reaction between photoacid and compound C) will substantially disappear.

10 Another advantage of the invention is, that irregularities that may occur in the process can be easily monitored and corrected. Sometimes formation of unwanted three dimensional structures occurs, due to side reactions, mistakes in the process, breakage of parts during build etc. The unwanted structures may float in the resin and may cause problems with building of new parts by interfering with the 15 stereolithography process. This problem is even more pronounced with filled resins since the density difference between resin and three dimensional object are rather low. The presence of color in the structures helps in identifying such pieces and in removing them from the vat/resin.

20 The three dimensional part made from the resin composition of the present invention will show a color. Preferably this color is uniform throughout the entire part. Different ways for measuring color exist. One may use absorbance measurements for transparent resins and/or parts, or color measurements for opaque and transparent resins and/or parts. For instance a transparent part preferably shows a maximum absorbance (in case the part is substantially transparent) of visible light of 25 at least 1.0 measured in the range from 400 to 650 nm on a sample having a thickness of 250 mil (or 0.635 cm). Preferably the maximum absorbance is more than 1.5. More preferably the maximum absorbance is more than 2.0.

30 Measurement of color can be performed with a Chroma meter. In case the resin composition and/or part are opaque due to for example the presence of a filler, the color of the resin and the part is measured with a Chroma meter on the part or resin as such. When the part or resin is transparent, measurement of the color is performed against a white background. A Chroma meter will give three values in the L*a*b color space (CIELAB 1976). The lightness (L) will be 100 for white materials and 0 for totally black materials. The 'a' and 'b' values represent the actual color: The '-a' 35 value represents green, '+a' represents red, '-b' represents blue and '+b' is yellow.

The 'a' value is between -60 and +60, 'b' is between -60 and + 60 . Parts having an 'a' and 'b' value between -20 and 20 will have a rather grey appearance. Parts having 'a' and 'b' values between -20 and -60 or 20 and 60 will be more colorful. The conventional resin compositions with and without fillers but no component C) will show

5 large L values between 90 and 100. Parts made by UV-curing a resin of the present invention will show a different color than the resin. This different color may be expressed as a change in L value, 'a' value or 'b' value relative to the resin. In case the L value does not change much (for example the color changes from red to blue), the 'a' or 'b' value will change at least with 20 units, preferably with 30 units. In most cases

10 however, the L value of the part will change relative to the resin, so that cured parts will have L-values between 0 and 85, preferably between 20 and 75. The 'a' and/or 'b' value of the cured part may stay the same as the values of the resin, as long as the L-value changes. Preferably the 'a' and /or 'b' value of a part will change by at least 10 units after cure of the resin. For instance the 'a' and/or 'b' value will change by at least

15 20 units.

An alternative way of indicating the color change that occurs after curing the resin composition of the present invention is by using the ratio L_c/L_u , wherein L_c is L of the cured part, and L_u is L of the uncured resin. L_c/L_u is preferably <0.95, more preferably < 0.9 or < 0.85. L_c and L_u are measured on a sample having a thickness of 20 mil (0.5 mm) or more, with a white Leneta-card background.

The resin compositions of the present invention may have cationically curable components A), and/or radically curable components D) as well as cationic photoinitiators B) and/or radical photoinitiators E). In case the compositions of the invention contain a filler F), the resin may be based on cationically curable

25 components, radically curable components or mixtures of these components (so called hybrid systems). When the compositions do not contain fillers, the resin may be preferably based on cationically curable components like epoxy components or hybrid systems.

Component A is at least one epoxy compound or a mixture of

30 different epoxy compounds. Epoxy compounds are compounds that possess on average at least one 1,2-epoxide group in the molecule. By "epoxide" is meant the three-membered ring having a structure represented by



5 The epoxide-containing materials, also referred to as epoxy materials, are cationically curable, by which is meant that polymerization and/or crosslinking of the epoxy group may be initiated by cations. The materials can be monomers, oligomers or polymers and are sometimes referred to as "resins." Such materials may have an aliphatic, aromatic, cycloaliphatic, arylaliphatic or heterocyclic structure; they
10 may comprise epoxide groups as side groups, or the epoxy groups may form part of an alicyclic or heterocyclic ring system. Epoxy groups may also be bound to for example siloxane containing backbones. Epoxy resins of those types are generally known and are commercially available.

 The epoxide-containing material (A) may for instance comprise at
15 least one liquid component such that the combination of materials is a liquid. Thus, the epoxide-containing material can be a single liquid epoxy material, a combination of liquid epoxy materials, or a combination of liquid epoxy material(s) and solid epoxy material(s) which is soluble in the liquid.

 Examples of suitable epoxy materials include polyglycidyl and
20 poly(methylglycidyl) esters of polycarboxylic acids, poly(oxiranyl) ethers of polyethers or epoxidised unsaturated fatty acids. The polycarboxylic acid can be aliphatic, such as, for example, glutaric acid, adipic acid and the like; cycloaliphatic, such as, for example, tetrahydrophthalic acid; or aromatic, such as, for example, phthalic acid, isophthalic acid, trimellitic acid, or pyromellitic acid. The polyether can be
25 poly(tetramethylene oxide). It is likewise possible to use carboxyterminated adducts, for example, of trimellitic acid and polyols, such as, for example, glycerol or 2,2-bis(4-hydroxycyclohexyl)propane. Suitable epoxidised unsaturated fatty acids may be obtained from for example linseed oil or perilla oil.

 Suitable epoxy materials also include polyglycidyl or poly(-methylglycidyl) ethers obtainable by the reaction of a compound having at least one free alcoholic hydroxy groups and/or phenolic hydroxy groups and a suitably substituted epichlorohydrin. The alcohols can be acyclic alcohols, such as, for example, ethylene glycol, diethylene glycol, and higher poly(oxyethylene) glycols; cycloaliphatic, such as, for example, 1,3- or 1,4-dihydroxycyclohexane, bis(4-hydroxycyclohexyl)methane, 2,2-bis(4-hydroxycyclohexyl)propane, or 1,1-bis(hydroxymethyl)cyclohex-3-ene; or contain aromatic nuclei, such as N,N-bis(2-

hydroxyethyl)aniline or p,p'-bis(2-hydroxyethylamino)diphenylmethane.

The epoxy compounds may also be derived from mono nuclear phenols, such as, for example, from resorcinol or hydroquinone, or they may be based on polynuclear phenols, such as, for example, bis(4-hydroxyphenyl)methane

5 (bisphenol F), 2,2-bis(4-hydroxyphenyl)propane (bisphenol A), or on condensation products, obtained under acidic conditions, of phenols or cresols with formaldehyde, such as phenol novolacs and cresol novolacs.

Suitable epoxy materials also include poly(N-glycidyl) compounds which are, for example, obtainable by dehydrochlorination of the reaction products of 10 epichlorohydrin with amines that comprise at least two amine hydrogen atoms, such as, for example, n-butylamine, aniline, toluidine, m-xylylene diamine, bis(4-aminophenyl)methane or bis(4-methylaminophenyl)methane. The poly(N-glycidyl) compounds also include, however, N,N'-diglycidyl derivatives of cycloalkyleneureas, such as ethyleneurea or 1,3-propyleneurea, and N,N'-diglycidyl derivatives of 15 hydantoins, such as of 5,5-dimethylhydantoin.

Examples of suitable epoxy materials include poly(S-glycidyl) compounds which are di-S-glycidyl derivatives which are derived from dithiols, such as, for example, ethane-1,2-dithiol or bis(4-mercaptomethylphenyl) ether.

Further examples of epoxide-containing materials are bis(2,3-20 epoxycyclopentyl)ether, 2,3-epoxy cyclopentyl glycidyl ether, 1,2-bis(2,3-epoxycyclopentyloxy)ethane, bis(4-hydroxycyclohexyl)methane diglycidyl ether, 2,2-bis(4-hydroxycyclohexyl)propane diglycidyl ether, 3,4-epoxycyclohexylmethyl-3,4-epoxycyclohexane, 3,4-epoxy-6-methylcyclohexylmethyl-3,4-epoxy-6-methylcyclohexanecarboxylate, di(3,4-epoxycyclohexylmethyl)hexanedioate, di(3,4-epoxy-6-methylcyclohexylmethyl)hexanedioate, ethylenebis(3,4-epoxycyclohexanecarboxylate), ethanediole(3,4-epoxycyclohexylmethyl)ether, vinylcyclohexene dioxide, dicyclopentadiene diepoxyde, α -(oxiranylmethyl)- ω -(oxiranylmethoxy) poly(oxy-1,4-butanediyl), diglycidyl ether of neopentyl glycol, or 2-(3,4-epoxycyclohexyl-5,5-spiro-3,4-epoxy)cyclohexane-1,3-dioxane, and combinations 25 thereof.

It is, however, also possible to use epoxy resins in which the 1,2-epoxy groups are bonded to different heteroatoms or functional groups. Those compounds include, for example, the N,N,O-triglycidyl derivative of 4-aminophenol, the glycidyl ether glycidyl ester of salicylic acid, N-glycidyl-N'-(2-glycidyloxypropyl)-5,5-dimethylhydantoin, or 2-glycidyloxy-1,3-bis(5,5-dimethyl-1-glycidylhydantoin-3-

yl)propane.

In addition, prereacted adducts of such epoxy resins with hardeners are suitable for epoxy resins.

It is of course also possible to use mixtures of epoxy materials in the 5 compositions according to the invention.

Preferred epoxy materials A) contain cycloaliphatic diepoxides.

Especially preferred cycloaliphatic diepoxides are bis(4-hydroxycyclohexyl)methane diglycidyl ether, 2,2-bis(4-hydroxycyclohexyl)propane diglycidyl ether, 3,4-epoxycyclohexylmethyl-3,4-epoxycyclohexanecarboxylate, 3,4-epoxy-6-10 methylcyclohexylmethyl-3,4-epoxy-6-methylcyclohexanecarboxylate, di(3,4-epoxycyclohexylmethyl)hexanedioate, di(3,4-epoxy-6-methylcyclohexylmethyl)hexanedioate, ethylenebis(3,4-epoxycyclohexanecarboxylate), ethanedioli(3,4-epoxycyclohexylmethyl) ether, 2-(3,4-epoxycyclohexyl-5,5-spiro-3,4-epoxy)cyclohexane-1,3-dioxane, and combinations thereof.

15 The epoxy materials can have molecular weights which vary over a wide range. In general, the epoxy equivalent weight, i.e., the number average molecular weight divided by the number of reactive epoxy groups, is preferably in the range of 60 to 1000.

Preferably the composition of the invention comprises from 30 to 20 80% by weight of the epoxide containing material A). Weight percentages of components of the resin composition of the present invention are related to the total weight of the radiation curable components of the composition, unless specified otherwise.

Component B) of the present resin composition comprises a cationic 25 photoinitiator. In the compositions according to the invention, any type of photoinitiator that, upon exposure to actinic radiation, forms cations that initiate the reactions of the epoxy material(s) can be used. There are a large number of known and technically proven cationic photoinitiators for epoxy resins that are suitable. They include, for example, onium salts with anions of weak nucleophilicity. Examples are halonium salts, 30 iodosyl salts or sulfonium salts, such as are described in published European patent application EP 153904 and WO 98/28663, sulfoxonium salts, such as described, for example, in published European patent applications EP 35969, 44274, 54509, and 164314, or diazonium salts, such as described, for example, in U.S. Patents 3,708,296 and 5,002,856. Other cationic photoinitiators are metallocene salts, such as described, 35 for example, in published European applications EP 94914 and 94915.

A survey of other current onium salt initiators and/or metallocene salts can be found in "UV Curing, Science and Technology", (Editor S. P. Pappas, Technology Marketing Corp., 642 Westover Road, Stamford, Conn., U.S.A.) Or "Chemistry & Technology of UV & EB Formulation for Coatings, Inks & Paints", Vol. 3 5 (edited by P. K. T. Oldring).

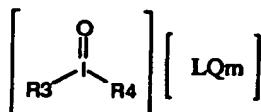
Preferred cationic photoinitiators are compounds of formula I, II or III below,

(I)

10

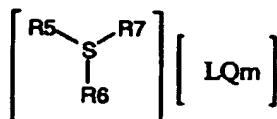


(II)



(III)

15



wherein:

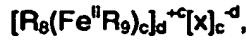
R₁, R₂, R₃, R₄, R₅, R₆, and R₇ are each independently of the others C6-C18 aryl-group that may be unsubstituted or substituted by suitable radicals, 20 L is boron, phosphorus, arsenic, or antimony, Q is a halogen atom or some of the radicals Q in an anion LQm⁻ may also be hydroxy groups, and m is an integer that corresponds to the valence of L plus 1. Examples of C6-C18 aryl are phenyl, naphthyl, anthryl, and 25 phenanthryl. Any substituents present for suitable radicals are alkyl, preferably C1-C6 alkyl, such as methyl, ethyl, n-propyl, isopropyl, n-butyl, sec-butyl, iso-butyl, tert-butyl, or the various pentyl or hexyl isomers, alkoxy, preferably C1-C6 alkoxy such as methoxy, ethoxy, propoxy, butoxy, pentyloxy, or hexyloxy, alkythio, preferably C1-C6

alkylthio, such as methylthio, ethylthio, propylthio, butylthio, pentylthio, or hexylthio, halogen, such as fluorine, chlorine, bromine, or iodine, amino groups, cyano groups, nitro groups, or arylthio, such as phenylthio.

5 Examples of preferred halogen atoms Q are chlorine and especially fluorine. Preferred anions LQ_m^- are BF_4^- , PF_6^- , AsF_6^- , SbF_6^- , and $SbF_5(OH)^-$.

Especially preferred are compositions comprising as the cationic photoinitiator a compound of formula III wherein R₅, R₆ and R₇ are aryl, aryl being especially phenyl or biphenyl, or mixtures of those two compounds.

10 Also preferred are compositions comprising as component B) a compound of formula (IV)



wherein,

c is 1 or 2,

15 d is 1,2,3,4 or 5,

X is a non-nucleophilic anion, especially PF_6^- , AsF_6^- , SbF_6^- , $CF_3SO_3^-$, $C_2F_5SO_3^-$, n- $C_3F_7SO_3^-$, n- $C_4F_9SO_3^-$, n- $C_6F_{13}SO_3^-$, or n- $C_8F_{17}SO_3^-$, R8 is a pi-arene, and R9 is an anion of a pi-arene, especially a cyclopentadienyl anion.

20 Examples of pi-arenes as R8 and anions of pi-arenes as R9 are to be found in published European patent application EP 94915.

Examples of preferred pi-arenes as R8 are toluene, xylene, ethylbenzene, cumene, methoxybenzene, methylnaphthalene, pyrene, perylene, stilbene, diphenylene oxide and diphenylene sulfide. Especially preferred are cumene, methylnaphthalene, or stilbene.

25 Examples of nonnucleophilic anions X⁻ are FSO_3^- , anions of organic sulfonic acids, of carboxylic acids, or anions LQ_m^- , as already defined above.

Preferred anions are derived from partially fluoro or perfluoroaliphatic or partially fluoro or perfluoro aromatic carboxylic acids, or especially from partially fluoro or perfluoroaliphatic or partially fluoro or perfluoroaromatic organic sulfonic acids, 30 or they are preferably anions LQ_m^- .

Examples of anions X⁻ are BF_4^- , PF_6^- , AsF_6^- , SbF_6^- , $SbF_5(OH)^-$, $CF_3SO_3^-$, $C_2F_5SO_3^-$, n- $C_3F_7SO_3^-$, n- $C_4F_9SO_3^-$, n- $C_6F_{13}SO_3^-$, n- $C_8F_{17}SO_3^-$, $C_6F_5SO_3^-$, phosphorus tungstate, or silicon tungstate. Preferred are PF_6^- , AsF_6^- , SbF_6^- , $CF_3S_3O^-$, $C_2F_5SO_3^-$, n- $C_3F_7SO_3^-$, n- $C_4F_9SO_3^-$, n- $C_6F_{13}SO_3^-$, and n- $C_8F_{17}SO_3^-$.

The metallocene salts can also be used in combination with oxidizing agents. Such combinations are described in published European patent application EP 126712.

The compositions of the invention preferably comprise from about 0.2

5 to about 10% by weight of cationic photoinitiator, based on the total weight of the composition.

In order to increase the light efficiency, or to sensitize the cationic photoinitiator to specific wavelengths, such as for example specific laser wavelengths or a specific series of laser wavelengths, it is also possible, depending on the type of

10 initiator, to use sensitizers. Examples of sensitizers are polycyclic aromatic hydrocarbons or aromatic keto compounds. Specific examples of preferred sensitizers are mentioned in published European patent application EP 153904. Other preferred sensitizers are benzoperylene, 1,8-diphenyl-1,3,5,7-octatetraene, and 1,6-diphenyl-1,3,5-hexatriene as described in U.S. Patent 5,667,937. It will be recognized that an

15 additional factor in the choice of sensitizer is the nature and primary wavelength of the source of actinic radiation.

The latent coloring component C) which forms colour or changes colour on contact with a photochemically generated photoacid is preferably a triaryl methane-, diphenyl methane-thiazine-, spiro-, lactam-, fluoran or isobenzofuranone-based colour former. Examples of Triaryl methane-based colour formers include, 3,3-bis(p-dimethylaminophenyl)-6-dimethylaminophthalide, 3,3-bis(p-dimethylaminophenyl)phthalide, 3-(p-dimethylaminophenyl)-3-(1,2-dimethylindole-3-yl)phthalide, 3-(p-dimethylaminophenyl)-3-(2-methylindole-3-yl)phthalide, 3,3-bis(1,2-dimethylindole-3-yl)-5-dimethylaminophthalide, 3,3-bis(1,2-dimethylindole-3-yl)-6-dimethylaminophthalide, 3,3-bis(9-ethylcarbazole-3-yl)-6-dimethylaminophthalide, 3,3-bis(2-phenylindole-3-yl)-6-dimethylaminophthalide, 3-p-dimethylaminophenyl-3-(1-methylpyrrole-3-yl)-6-dimethylaminophthalide, etc., especially triphenyl methanes e.g., Crystal Violet Lactone.

Diphenylmethane-based colour formers include 4,4'-bis-

30 dimethylaminobenzhydryl benzyl ether, N-halophenyl-leucoauramine and N-2,4,5-trichlorophenyl-leucoauramine.

Thiazine-based colour formers include benzoyl-leucomethylene blue and p-nitrobenzoyl-leucomethylene blue.

Spiro-based colour formers include 3-methyl-spiro-dinaphthopyran, 3-

35 ethyl-spiro-dinaphthopyran, 3-phenyl-spirodinaphthopyran, 3-benzyl-spiro-

dinaphthopyran, 3-methyl-naphtho-(6'-methoxybenzo)spiropyran and 3-propyl-spiro-dibenzopyran.

Lactam-based colour formers include rhodamine-b-anilinolactam, rhodamine-(p-nitroanilino)lactam and rhodamine-(o-chloroanilino)lactam.

5 Fluoran-based colour formers include 3,6-dimethoxyfluoran, 3,6-dieethoxyfluoran, 3,6-dibutoxyfluoran, 3-dimethylamino-7-methoxyfluoran, 3-dimethylamino-6-methoxyfluoran, 3-dimethylamino-7-methoxyfluoran, 3-diethylamino-7-chlorofluoran, 3-diethylamino-6-methyl-7-chlorofluoran, 3-diethylamino-6,7-dimethylfluoran, 3-(N-ethyl-p-toluidino)-7-methylfluoran, 3-diethylamino-7-(N-acetyl-N-methylamino)fluoran, 3-diethylamino-7-N-methylaminofluoran, 3-diethylamino-7-dibenzylaminofluoran, 3-diethylamino-5-methyl-7-dibenzylaminofluoran, 3-diethylamino-7-(N-methyl-N-benzylamino)fluoran, 3-diethylamino-7-(N-chloroethyl-N-methylamino)fluoran, 3-diethylamino-7-diethylaminofluoran, 3-(N-ethyl-p-toluidino)-6-methyl-7-phenylaminofluoran, 3-(N-ethyl-p-toluidino)-6-methyl-7-phenylaminofluoran, 15 3-diethylamino-7-(2-carbomethoxy-phenylamino)fluoran, 3-(N-ethyl-N-isoamylamino)-6-methyl-7-phenylaminofluoran, 3-(N-cyclohexyl-N-methylamino)-6-methyl-7-phenylaminofluoran, 3-pyrrolidino-6-methyl-7-phenylaminofluoran, 3-piperidino-6-methyl-7-phenylaminofluoran, 3-diethylamino-6-methyl-7-xylidinofluoran, 3-diethylamino-7-(o-chlorophenylamino)fluoran, 3-dibutylamino-7-(o-chlorophenylamino)fluoran and 3-pyrrolidino-6-methyl-7-p-butylphenylaminofluoran.

20

Colour formers permitting the production of a wide range of colours are known and have been described, for example, by Peter Gregory in High-Technology Applications of Organic Colourants, Plenum Press, pages 124-134.

Preferred coloring components are the isobenzofuranone-based

25 color formers and colour formers that are available under the tradenames of Copikem and Pergascript. Examples of these preferred coloring components are Copikem 20 (3,3-Bis (1-butyl-2-methyl-H-indol-3-yl) - 1-(3H)-isobenzofuranone), Copikem 5 (2'-Di(phenylmethyl) amino-6'-(diethylamino)spiro(isobenzofuran-1(3H),9'-(9H)xanthen)-3-one), Copikem 14 (a substituted phthalide), Copikem 7 (3-((4Dimethylamino)-phenyl)-3-(1-butyl-2methylindol-3yl) -6-dimethyamino) -1(3H)-isobenzofuranone), Copikem 37 (2-(2-Octoxyphenyl)-4-(4-dimethylaminophenyl)-6-(phenyl)pyridine), Pergascript Black I-R (6''-(Dimethylamino)-3''-methyl - 2'' - (phenylamino)spiro(isobenzofuran-1 (3H), 9'' (9H)xanthem-3-one), and Pergascript Color Formers (like Diamiofluoran compounds, Bisaryl carbazolyl methane compounds, Phthalide compounds, Bisindolyl phthalide compounds, Aminofluoran compounds and Quinazoline compounds).

30

35

The amount of component C) is preferably between 0.0001 and 1 wt%, more preferably between 0.0005 and 0.1 wt%.

A component D) may be present in the resin composition of the present invention. This component D) comprises a compound having at least one

- 5 ethylenically unsaturation which can be polymerized with radicals. Specific examples of suitable ethylenic unsaturation are groups containing acrylate, methacrylate, styrene, vinylether, vinyl ester, N-substituted acrylamide, N-vinyl amide functionalities or maleate esters, and fumarate esters. Specifically, the ethylenic unsaturation is provided by a group containing acrylate, methacrylate, N-vinyl, or styrene functionality. For
- 10 example, component D comprises one or more compounds having one or more (meth)acrylate functionalities.

The free-radical polymerizable acrylic materials that may be used in the composition are compounds that have, on average, at least one acrylic group which can be either the free acid or an ester. By "acrylic" is meant the group -

- 15 $\text{CH}=\text{CR1CO2R2}$, where R1 can be hydrogen or methyl and R2 can be hydrogen or alkyl. By "(meth)acrylate" is meant an acrylate, methacrylate or combinations thereof. The acrylic materials undergo polymerization and/or crosslinking reactions initiated by free radicals. The acrylic materials can be monomers, oligomers or polymers. It is preferred that the acrylic material be a monomer or oligomer.

- 20 Suitable as the acrylic component are, for example, the diacrylates of cycloaliphatic or aromatic diols, such as 1,4-dihydroxymethylcyclohexane, 2,2-bis(4-hydroxycyclohexyl)propane, 1,4-cyclohexanedimethanol, bis(4-hydroxycyclohexyl)methane, hydroquinone, 4,4-dihydroxybiphenyl, bisphenol A, bisphenol F, bisphenol S, ethoxylated or propoxylated bisphenol A, ethoxylated or 25 propoxylated bisphenol F, or ethoxylated or propoxylated bisphenol S, and combinations thereof. Such acrylates are known and some of them are commercially available.

- 30 Suitable as aromatic tri(meth)acrylates are, for example, the reaction products of triglycidyl ethers of trihydric phenols, and phenol or cresol novolacs having three hydroxy groups with (meth)acrylic acid. Preferably the acrylic material contains 1,4-dihydroxymethyl-cyclohexane diacrylate, bisphenol A diacrylate, ethoxylated bisphenol A diacrylate and combinations thereof.

- 35 Compositions wherein the acrylic component contains an acrylate of bisphenol A diepoxide such as Ebecryl 3700® from UCB Chemical Corporation, Smyrna, Georgia, a mixed acrylate/epoxy compound of bisphenol A such as Ebecryl

3605® or an acrylate of 1,4-cyclohexanedimethanol are preferred for compositions used in this invention.

In addition to or instead of the aromatic or cycloaliphatic acrylic material, other acrylic materials can be present. Poly(meth)acrylates having

5 functionality of greater than 2 may, where appropriate, be used in the compositions according to the invention. These can be, for example, tri, tetra, or pentafunctional monomeric or oligomeric aliphatic (meth)acrylates.

Suitable as aliphatic polyfunctional (meth)acrylates are, for example, the triacrylates and trimethacrylates of hexane-2,4,6-triol, glycerol, or 1,1,1-

10 trimethylopropane, ethoxylated or propoxylated glycerol, or 1,1,1-trimethylopropane and the hydroxy group-containing tri(meth)acrylates which are obtained by the reaction of triepoxy compounds, such as, for example, the triglycidyl ethers of the mentioned triols, with (meth)acrylic acid. It is also possible to use, for example, pentaerythritol tetra-acrylate, bis(trimethylopropane) tetra-acrylate, pentaerythritol

15 monohydroxytri(meth)acrylate, or dipentaerythritol monohydroxypenta(meth)acrylate, and combinations thereof.

It is also possible to use hexafunctional urethane (meth)acrylates.

Those urethane (meth)acrylates are known to the person skilled in the art and can be prepared in known manner, for example by reacting a hydroxy-terminated polyurethane

20 with acrylic acid or methacrylic acid, or by reacting an isocyanate-terminated prepolymer with hydroxyalkyl (meth)acrylates to follow the urethane (meth)acrylate. Also useful are acrylates and methacrylates such as tris(2-hydroxyethyl)isocyanurate triacrylate.

The amount of component D) is for example between 5 and 60 wt%,

25 preferably between 10 and 40 wt%.

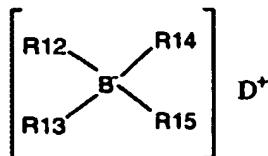
In the compositions according to the invention a radical photoinitiator E) may be used, especially in combination with component D). Any type of photoinitiator that forms free radicals when the appropriate irradiation takes place can be used. Typical compounds of known photoinitiators are benzoins, such as benzoin,

30 benzoin ethers, such as benzoin methyl ether, benzoin ethyl ether, and benzoin isopropyl ether, benzoin phenyl ether, and benzoin acetate, acetophenones, such as acetophenone, 2,2-dimethoxyacetophenone, 4-(phenylthio)acetophenone, and 1,1-dichloroacetophenone, benzil, benzil ketals, such as benzil dimethyl ketal, and benzil diethyl ketal, anthraquinones, such as 2-methylanthraquinone, 2-ethylanthraquinone, 2-tertbutylanthraquinone, 1-chloroanthraquinone, and 2-aminanthraquinone, also

triphosphine, benzoylphosphine oxides, such as, for example, 2,4,6-trimethylbenzoyldiphenylphosphine oxide (Lucirin TPO), benzophenones, such as benzophenone, and 4,4'-bis(N,N'-dimethylamino)benzophenone, thioxanthones and xanthones, acridine derivatives, phenazene derivatives, quinoxaline derivatives or I-phenyl-1,2-propanedione-2-O-benzoyloxime, I-aminophenyl ketones or I-hydroxyphenyl ketones, such as I-hydroxycyclohexyl phenyl ketone, phenyl (1-hydroxyisopropyl)ketone and 4-isopropylphenyl(1-hydroxyisopropyl)ketone, or triazine compounds, for example, 4"-methyl thiophenyl-1-di(trichloromethyl)-3,5-S-triazine, S-triazine-2-(stilbene)-4,6-bistrichloromethyl, and paramethoxy styryl triazine, all of which are known compounds.

Especially suitable free-radical photoinitiators, which are normally used in combination with a He/Cd laser, operating at for example 325 nm, an Argon-ion laser, operating at for example 351 nm, or 351 and 364 nm, or 333, 351, and 364 nm, or a frequency tripled YAG solid state laser, having an output of 351 or 355 nm, as the radiation source, are acetophenones, such as 2,2-dialkoxybenzophenones and 1-hydroxyphenyl ketones, for example 1-hydroxycyclohexyl phenyl ketone, 2-hydroxy-1-[4-(2-hydroxyethoxy)phenyl]-2-methyl-1-propanone, or 2-hydroxyisopropyl phenyl ketone (also called 2-hydroxy-2,2-dimethylacetophenone), but especially 1-hydroxycyclohexyl phenyl ketone. Another class of free-radical photoinitiators comprises the benzil ketals, such as, for example, benzil dimethyl ketal. Especially an alpha-hydroxyphenyl ketone, benzil dimethyl ketal, or 2,4,6-trimethylbenzoyldiphenylphosphine oxide is used as photoinitiator.

Another class of suitable free radical photoinitiators comprises the ionic dye-counter ion compounds, which are capable of absorbing actinic rays and producing free radicals, which can initiate the polymerization of the acrylates. The compositions according to the invention that comprise ionic dye-counter ion compounds can thus be cured in a more variable manner using visible light in an adjustable wavelength range of 400 to 700 nanometers. Ionic dye-counter ion compounds and their mode of action are known, for example from published European patent application EP 223587 and U.S. Patents 4,751,102, 4,772,530 and 4,772,541. There may be mentioned as examples of suitable ionic dye-counter ion compounds the anionic dye-iodonium ion complexes, the anionic dye-pyrilliun ion complexes and, especially, the cationic dye-borate anion compounds of the following formula



wherein D⁺ is a cationic dye and R₁₂, R₁₃, R₁₄, and R₁₅ are each independently of the others alkyl, aryl, alkaryl, allyl, aralkyl, alkenyl, alkynyl, an alicyclic or saturated or unsaturated heterocyclic group. Preferred definitions for the radicals R₁₂ to R₁₅ can be found, for example, in published European patent application EP 223587.

The compositions of the invention preferably comprise from about 0.01 to about 10% by weight of free-radical photoinitiator, based on the total weight of the composition.

The resin composition of the present invention may also contain a filler 10 F). The filler may be any substance without special limitation. An inorganic substance is preferred from the point of view of the water-resisting capabilities and mechanical properties of the parts made of the resin composition of the invention. Examples of inorganic fillers are silica, glass powder, alumina, alumina hydrate, magnesium oxide, magnesium hydroxide, barium sulfate, calcium sulfate, calcium carbonate, magnesium 15 carbonate, silicate mineral, diatomaceous earth, silica sand, silica powder, titanium oxide, aluminum powder, bronze, zinc powder, copper powder, lead powder, gold powder, silver dust, glass fiber, titanic acid potassium whiskers, carbon whiskers, sapphire whiskers, verification rear whiskers, boron carbide whiskers, silicon carbide whiskers, and silicon nitride whiskers.

20 The condition of the surface of the particles of the filler used and the impurities contained in filler from the manufacturing process can affect the curing reaction of the resin composition. In such cases, it is preferable to wash the filler particles or coat the particles with an appropriate primer as a method of improving the curing properties.

25 These inorganic fillers may also be surface-treated with a silane coupling agent. Silane coupling agents which can be used for this purpose include vinyl trichlorosilane, vinyl tris (β-methoxyethoxy) silane, vinyltriethoxy silane, vinyltrimethoxy silane, γ-(methacryloxypropyl)trimethoxy silane, β-(3,4-epoxycyclohexyl)ethyltrimethoxy silane, γ-glycydoxypropyltrimethoxy silane, γ-glycydoxypropylmethyl diethoxy silane, 30 N-β(aminoethyl)-γ-aminopropyltrimethoxy silane, N-β-(aminoethyl)-γ-

aminopropylmethyldimethoxy silane, γ -aminopropyltriethoxysilane, N-phenyl- γ -amino propyl trimethoxy silane, γ -mercaptopropyl trimethoxysilane, and γ -chloropropyltrimethoxy silane.

The above inorganic fillers may be used singly or in combination of
5 two or more.

The amount of filler may be present between 0 and 95 wt% relative to the total weight of the composition. For instance the filler may be present in 10-90 wt%, or for example between 20-80 wt%.

A hydroxyl-group containing material G) may be used in the present
10 invention. Component G may be any liquid organic material having hydroxyl functionality of at least 1, and preferably at least 2. The material may be a liquid or a solid that is soluble or dispersible in the remaining components. Preferably the material is substantially free of any groups which substantially slowdown the curing reaction.

Preferably the organic material contains two or more primary or
15 secondary aliphatic hydroxyl groups, by which is meant that the hydroxyl group is bonded directly to a non-aromatic carbon atom. Monomers, oligomers or polymers can be used. The hydroxyl equivalent weight, i.e., the number average molecular weight divided by the number of hydroxyl groups, is preferably in the range of 31 to 5000.

Representative examples of suitable organic materials having a
20 hydroxyl functionality of 1 include alkanols, monoalkyl ethers of polyoxyalkyleneglycols, monoalkyl ethers of alkylenglycols, and others, and combinations thereof.

Representative examples of useful monomeric polyhydroxy organic materials include alkylene and arylalkylene glycols and polyols, such as 1,2,4-butanetriol, 1,2,6-hexanetriol, 1,2,3-heptanetriol, 2,6-dimethyl-1,2,6-hexanetriol,
25 (2R,3R)-(-)-2-benzyloxy-1,3,4-butanetriol, 1,2,3-hexanetriol, 1,2,3-butanetriol, 3-methyl-1,3,5-pantanetriol, 1,2,3-cyclohexanetriol, 1,3,5-cyclohexanetriol, 3,7,11,15-tetramethyl-1,2,3-hexadecanetriol, 2-hydroxymethyltetrahydropyran-3,4,5-triol, 2,2,4,4-tetramethyl-1,3-cyclobutanediol, 1,3-cyclopentanediol, trans-1,2-cyclooctanediol, 1,16-hexadecanediol, 3,6-dithia-1,8-octanediol, 2-butyne-1,4-diol, 1,3-propanediol, 1,4-
30 butanediol, 1,5-pantanediol, 1,6-hexanediol, 1,7-heptanediol, 1,8-octanediol, 1,9-nonanediol, 1-phenyl-1,2-ethanediol, 1,2-cyclohexanediol, 1,5-decalindiol, 2,5-dimethyl-3-hexyne-2,5-diol, 2,7-dimethyl-3,5-octadiyne-2-7-diol, 2,3-butanediol, 1,4-cyclohexanediethanol, and combinations thereof.

Representative examples of useful oligomeric and polymeric hydroxyl-containing materials include polyoxyethylene and polyoxypropylene glycols and triols of molecular weights from about 200 to about 10,000; polytetramethylene glycols of varying molecular weight; copolymers containing pendant hydroxy groups

5 formed by hydrolysis or partial hydrolysis of vinyl acetate copolymers, polyvinylacetal resins containing pendant hydroxyl groups; hydroxy-terminated polyesters and hydroxy-terminated polylactones; hydroxy-functionalized polyalkadienes, such as polybutadiene; aliphatic polycarbonate polyols, such as an aliphatic polycarbonate diol; and hydroxy-terminated polyethers, and combinations thereof.

10 Preferred hydroxyl-containing monomers are 1,4-cyclohexanedimethanol and aliphatic and cycloaliphatic monohydroxy alkanols.

Preferred hydroxyl-containing oligomers and polymers include hydroxyl and hydroxyl/epoxy functionalized polybutadiene, 1,4-cyclohexanedimethanol, polycaprolactone diols and triols, ethylene/butylene polyols, and monohydroxyl

15 functional monomers. Preferred examples of polyether polyols are polypropylene glycols of various molecular weights and glycerol propoxylate-B-ethoxylate triol. Especially preferred are linear and branched polytetrahydrofuran polyether polyols available in various molecular weights, such as for example 250, 650, 1000, 2000, and 2900 MW.

20 The composition of the present invention may contain preferably up to 30 wt% of polyol, preferably between 3 and 20 weight percent.

Preferably the resin composition of the present invention is a hybrid system, which means that the resin composition contains both radically and cationically polymerizable components and also radical and cationic photoinitiators.

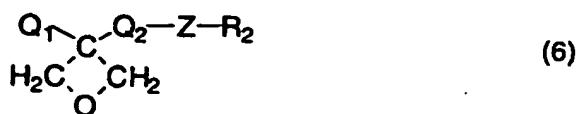
25 The resin composition of the present invention may comprise other cationically curable components like for example a cyclic ether component, vinyl ether component, cyclic lactone component, cyclic acetal component, cyclic thioether component, spiro orthoester component, and/or oxetane-functional component.

Oxetanes are components comprising one or more oxetane groups, 30 i.e. one or more four-member ring structures according to formula (5):



(5)

Examples of oxetanes include components represented by the following formula (6):



5 wherein

Q, represents a hydrogen atom, an alkyl group having 1 to 6 carbon atoms (such as a methyl, ethyl, propyl, or butyl group), a fluoroalkyl group having 1 to 6 carbon atoms, an allyl group, an aryl group, a furyl group, or a thienyl group;

Q₂ represents an alkylene group having 1 to 6 carbon atoms (such as a methylene, ethylene, propylene, butylene, pentylene, or hexylene group).

10 ethylene, propylene, or butylene group), or an alkylene group containing an ether linkage, for example, an oxyalkylene group, such as an oxyethylene, oxypropylene, or oxybutylene group

Z represents an oxygen atom or a sulphur atom; and

15 R_2 represents a hydrogen atom, an alkyl group having 1-6 carbon atoms (e.g. a methyl group, ethyl group, propyl group, or butyl group), an alkenyl group having 2-6 carbon atoms (e.g. a 1-propenyl group, 2-propenyl group, 2-methyl-1-propenyl group, 2-methyl-2-propenyl group, 1-butenyl group, 2-butenyl group, or 3-butenyl group), an aryl group having 6-18 carbon atoms (e.g. a phenyl group, naphthyl group, anthranyl group, or phenanthryl group), a substituted or unsubstituted aralkyl group having 7-18 carbon atoms (e.g. a benzyl group, fluorobenzyl group, methoxy benzyl group, phenethyl group, styryl group, cinnamyl group, ethoxybenzyl group), an aryloxyalkyl group (e.g. a phenoxyethyl group or phenoxyethyl group), an alkylcarbonyl group having 2-6 carbon atoms (e.g. an ethylcarbonyl group, propylcarbonyl group, or butylcarbonyl group), an alkoxy carbonyl group having 2-6 carbon atoms (e.g. an ethoxycarbonyl group, propoxycarbonyl group, or butoxycarbonyl group), an N-alkylcarbamoyl group having 2-6 carbon atoms (e.g. an ethylcarbamoyl group, propylcarbamoyl group, butylcarbamoyl group, or pentylcarbamoyl group), or a polyether group having 2-1000 carbon atoms.

20

25

The process for producing three-dimensional articles from the compositions of the invention, as discussed above, generally involves exposure of successive thin layers of the liquid composition to actinic radiation. A thin layer of the photosensitive composition of the invention is coated onto a surface. This is most conveniently done if the composition is a liquid. However, a solid composition can be melted to form a layer. Also a viscosity reducible composition maybe used which may be applied under shear. After coating of a thin layer, this layer is then exposed

imagingwise to actinic radiation to form a first imaged cross-section. The radiation must provide sufficient exposure to cause substantial curing of the photosensitive composition in the exposed areas. By "substantial curing" it is meant that the photosensitive composition has reacted to an extent such that the exposed areas are

5 physically differentiable from the unexposed areas. For liquid, gel, paste or semi-solid photosensitive compositions, the cured areas will have hardened or solidified to a non-fluid form. For solid photosensitive compositions, the exposed areas will have a higher melting point than the non-exposed areas. Preferably, the exposure is such that portions of each successive layer are adhered to a portion of a previously exposed

10 layer or support region, or to portions of a platform surface. An additional (second) thin layer of photosensitive composition is then coated onto the first imaged cross-section and imagingwise exposed to actinic radiation to form an additional (second) imaged cross-section. These steps are repeated with the "nth" thin layer of photosensitive composition being coated onto the "(n-1)th" imaged cross-section and exposing to

15 actinic radiation. The repetitions are carried out a sufficient number of times to build up the entire three-dimensional article.

The radiation is preferably in the range of 280-650 nm. Any convenient source of actinic radiation can be used, but lasers are particularly suitable. Useful lasers include HeCd, argon, nitrogen, metal vapor, and NdYAG lasers. The

20 exposure energy is for example in the range of 10-500 mJ/cm², preferably in the range between 20 and 250 mJ/cm², for instance between 30 and 150 mJ/cm². Suitable methods and apparatus for carrying out the exposure and production of three-dimensional articles have been described in, for example, U.S. Patents 4,987,044, 5,014,207, and 5,474,719, which teaches the use of pseudoplastic, plastic flow,

25 thixotropic, gel, semi-solid and solid photopolymer materials in the solid imaging process. These US-patents are incorporated herein by reference.

In general, the three-dimensional article formed by exposure to actinic radiation, as discussed above, is not fully cured, by which is meant that not all of the reactive material in the composition has reacted. Therefore, there is often an additional

30 step of postcuring the article. This can be accomplished by further irradiating with actinic radiation, heating, or both. Exposure to actinic radiation can be accomplished with any convenient radiation source, generally a UV light, for a time ranging from about 10 to over 60 minutes. Heating is generally carried out at a temperature in the range of about 75-200 °C, for a time ranging from about 10 to over 60 minutes.

35 It has been surprisingly found, that thermal postcure may have an

influence on the color of the cured part. Thermal postcure may 'bleach' the article giving an article having a higher L value than the article showed after preparation with for example a laserbeam. This gives the opportunity to maintain the advantages of cleaning a part made from for example a filled resin, but gives the flexibility to choose 5 intensity of color upon request of the customer by simple postcure methods. The 'a' and 'b' values may be reduced to values closer to zero. Bleaching takes place for example when thermal postcure is performed at high temperatures for prolonged periods of time, like for example 2 hours postcure at 160 °C.

10 **Experimental part**

Color measurement is performed with a Macbeth Color Eye 7000 spectrophotometer, or equivalent, whereby the instrument is calibrated using the white tile and the black trap (zero reflectance) with Large Area View (LAV) according to the manufacturer's instructions. The measurement conditions are selected as follows:

15 Large Area View (LAV) and Specular Component Included (SCI), D65 illuminant and 10° observer. Color measurements can be performed on films or flat surfaces with the Macbeth Color Eye 7000, or also with manual chromameters which are commercially available. The films have been placed against a white background of a Leneta card, having a background color of L 94.5, a = -0.9 and b = 4.7.

20 Samples have been prepared by taking resin from a container after homogenizing the resin in the container by mechanical stirring. Samples for color measurement have been prepared of cured ((semi) transparent) films, paste resins with fillers and parts made from the paste resins. Sample of transparent resin have been prepared by making a resin film of 20 mils (0.51 mm) thickness and curing the resin 25 with a D-lamp using two doses radiation of 0.13 J/cm² under a nitrogen atmosphere. Samples of filled resin have been prepared by making a 6 mils (0.15 mm) film using 200 mJ/cm² radiation.

Absorbance of transparent liquid resins has been measured by putting a resin in a 1 cm quartz cuvet and measuring the UV-VIS spectrum between 30 400 and 650 nm in a standard spectrophotometer. The highest absorbance in that wavelength range is taken as the absorbance value. Absorbance of a cured transparent part is measured on a part that has been made on a standard SLA machine equipped with an Argon-Ion laser. The exposure given was 85 mJ/cm², to build a part having a thickness of 250 mil (0.635 cm) with layers of 6 mil (0.15 mm).

Examples

A commercially available compound SOMOS ® 7720 was taken as a starting epoxy based resin composition. Somos ® 7720 contains between 40 and 80 wt% of epoxy compounds, between 3 and 15 wt% polyols, between 10 and 40 wt% of 5 acrylates, radical and cationic photoinitiators.

Various Copikem latent coloring components C) were added to the base Somos 7720 composition in different amounts. The absorbance of Somos 7720 with or without Copikem latent coloring components is the same: the absorbance in the visible region between 400 and 650 nm is lower than 10 0.2, measured on a 1 cm thick liquid sample of the resin composition. The absorbance in the region between 450 and 700 nm is even below 0.10. All resin compositions did not show any color except the natural slight yellow color, which may be present in these types of resin compositions.

Absorbance has been measured by making a standard UV-VIS 15 spectrum in the region between 400 and 650 nm on a cured sample having a thickness of 250 mil (0.635 cm). The maximum absorbance is the highest absorbance measured in the spectrum.

Component C (wt%)	7720	Ex 1	Ex 2	Ex 3	Ex 4	Ex 5	Ex 6
Copikem 20-magenta	-	0.0018	0.003	-	-	0.0014	0.0014
Copikem-4 black	-	-	-	0.0085	-	0.0085	-
Copikem -14 orange	-	-	-	-	0.0085	0.0030	0.0030
Pergascript Black I-R	-	-	-	-	-	-	0.0085
Max absorbance	0.27	0.76	1.3	2.31	1.44	2.65	3.3
At wavelength (nm)	400	530	530	580	450	450	450

20 Copikem 20 is (3,3-Bis (1-butyl -2-methyl-H-indol-3-yl) - 1-(3H)-isobenzofuranone), Copikem 4 (CAS No. 29512-49-0) is 6'-(diethylamino)-3'-methyl-2'- (phenylamino)spiro(isobenzofuran-1(3H0,9'-(9H)xanthen-3-one.

Copikem 14 (Orange) (CAS No. 67697-15-0) is a substituted phthalide
Pergascript Black I-R is (6''-(Dimethylamino)-3''-methyl - 2'' -

25 (phenylamino)spiro(isobenzofuran-1 (3H), 9'' (9H)xanthem-3-one).

All examples show that addition of component C gives objects having a color, which color is different than the original natural color of the resin composition.

Addition of higher amounts of coloring agent (examples 1 and 2) gives a higher

maximum absorbance. Examples 2 to 4 show that different types of coloring agents may be used. Example 5 and 6 show that also combination of coloring agents is possible. All parts show a uniform coloring. The color forms during the (first) exposure of the resin to irradiation. Further post treatment of the part with UV does not 5 substantially change the color and/or the absorbance of the part.

Surprisingly, the physical properties of the colored parts are substantially the same as the part made from the resin having no component C. The part made in example 6 (with a UV-postcuring treatment) has the following properties (properties from 7720 example without coloring component in brackets):

10 Flexural Modulus data: Young's modulus 230 (236), Flex strength KSI 7.00 (7.8), % strain 4.35 (5.1).

Example 7

15 The resin of example 6 and the commercial Somos 7720, having no latent coloring component C, have been cured under nitrogen with two doses of 0.13 J/cm² radiation of a D lamp into a 20 mil (0.51 mm) thick film. Color measurement has been performed with a Macbeth Color Eye 7000 spectrophotometer. Results are shown in table 2.

	7720 film	Example 6 resin film
L	91.3	77
'a'	-2	0
'b'	8	10

20 Table 2

Examples 8-10

A base resin containing 4.1 wt% 3,4-Epoxy Cyclohexyl Methyl-3,4-Epoxy Cyclohexyl Carboxylate (Uvacure 1500), 6.6 wt% 3-Ethyl-3-(hydroxymethyl)oxetane (UVR6000), 16.4 wt% Epon 825, 4 wt% Dipentaerithritol pentaacrylate (Sartomer SR-399), 63 wt% silica (Siltex 44), 1.3 wt% Triton X-100, 2.9 wt% Thixatrol XT, 0.23 wt% 1-Hydroxycyclohexyl phenyl ketone (Irgacure 184), and 1.3 wt% CPI-6976 (a mixture of Sulfonium(thiodi-4,1-phenylene)bis[diphenyl-bis[(OC-6-11)hexafluoroantimonate(1-)]]) and p-Thiophenoxyphenyldiphenylsulfonium) is mixed 30 with different amounts of latent coloring component C (Copikern 20 Magenta).

6 mils films were prepared by applying a 6 mil film onto a petridish, irradiating it with an solid state laser, having a wavelength of 365 nm, with an exposure of 200 mJ/cm². The film was cleared with tri(propylene glycol) methyl ether and rinsed with isopropylalcohol before postcure. Two thin films for each sample were postcured in a postcuring

5 apparatus for 15 minutes on each side of the film. The postcuring apparatus (3D systems) contains a 10 bulb unit having Philips TLK/05 40W bulbs.

Two other thin films for each sample were thermal postcured in an oven for two hours at 160 °C, with a warming up time of 2 hours and a cooling down time of 2 hours. The color is measured both of the resins before cure and of the parts made after

10 photofabrication and either UV-postcure or thermal postcure.

The uncured resin has the following color characteristics L = 92, a = -0.8, b = 5.

Example	Wt% dye	Object after UV-cure and UV postcure				Object after UV-cure and thermal postcure			
		L	Lc/Lu	A	B	L	Lc/Lu	a	b
A	0	92	1	-0.8	6	92	1	-0.8	6
8	0.00053	81	0.88	13	-3.3	87	0.95	-0.5	8.7
9	0.00264	73	0.79	34	-15	87	0.95	3.1	4.8
10	0.01190	60	0.65	48	-20	78	0.85	15	-3.1

CLAIMS

1. A radiation curable resin composition suitable for making three dimensional objects comprising
 - 5 A) at least one epoxy compound
 - B) a cationic photoinitiator,
wherein the resin composition has a first color or no color before cure and wherein a three dimensional object made from the resin by subjecting the resin to radiation shows a second color which is different from the color of the resin composition before cure.
- 10 2. A radiation curable resin composition suitable for making three dimensional objects comprising
 - A) at least one epoxy compound
 - B) a cationic photoinitiator and
 - 15 C) a latent coloring component,
wherein the resin composition shows
 - i an absorbance of visible light in the region between 400 and 650 nm of less than 0.3 measured on a sample having a thickness of 1 cm; or
 - ii a L-value between 90 and 100 in the CIELAB L*a*b color space,
20 measured on a 0.5 mm sample against a white Leneta card background.
 3. The resin composition according to claim 2, wherein the absorbance of visible light of the resin is less than 0.2.
 4. The resin composition according to anyone of claims 1-3, wherein a three dimensional object made from the resin composition by subjecting the resin to radiation shows color and has
 - 25 i a maximum absorbance of visible light of at least 1.0 measured in the range from 400 to 650 nm on a sample having a thickness of 250 mil (0.635 cm); or
 - ii a L-value between 0 and 85.
 - 30 5. The resin composition according to claim 4, wherein the maximum absorbance is more than 1.5, or wherein the L-value is between 20 and 75.
 6. The resin composition according to anyone of claims 1-5, wherein the difference between the 'a' and/or 'b' values of the object and the resin are larger than 20 units, wherein 'a' and 'b' relate to the L*a*b color space.
 - 35 7. A radiation curable composition suitable for making three dimensional objects

comprising a radiation curable component, a photoinitiator and a filler, wherein the resin composition has a first color or no color before cure and wherein a three dimensional object made from the resin by subjecting the resin to radiation shows a second color which is different from the color of the resin.

- 5 8. The resin composition according to claim 7, wherein the resin composition has a L-value between 90 and 100, and wherein the object has a L value between 0 and 85.
9. The resin composition according to claims 1, 7 or 8, wherein the resin composition contains a latent coloring component C).
- 10 10. The resin composition according to claims 2-5, 9, wherein the component C is an isobenzofuranone based colour former.
11. The resin composition according to anyone of claims 2-5, 9-10, wherein the amount of component C is between 0.0001 and 1 wt%.
12. The resin composition according to anyone of the preceding claims, wherein a component A is present that contains a cycloaliphatic diepoxide.
- 15 13. The resin composition according to anyone of the preceding claims, wherein a component D) is present which contains an ethylenically unsaturation
14. The resin composition according to claim 13, wherein component D) comprises a (meth)acrylate functionality.
- 20 15. The resin composition according to claim 14, wherein component D comprises a mono or di acrylate of bisphenol A diepoxide.
16. The resin composition according to anyone of the preceding claims, wherein the resin composition contains a radical photoinitiator E).
17. The resin composition according to anyone of the preceding claims, wherein 25 the resin composition contains at least two epoxy compounds.
18. The resin composition according to anyone of the preceding claims, wherein the resin composition contains a polyol F).
19. A resin composition suitable for making three dimensional objects comprising 30 A) 30-80 wt% of at least one epoxy compound
B) a cationic photoinitiator,
C) a latent coloring component
D) a (meth)acrylate component
E) a radical photoinitiator
F) a polyol
- 35 wherein the resin composition has a first color or no color before cure and

wherein a three dimensional object made from the resin by subjecting the resin to radiation shows a second color which is different from the color of the resin composition before cure.

20. A process for forming a three-dimensional article comprising:
 - 5 (1) coating a layer of a composition onto a surface, wherein the composition is used as defined in anyone of claims 1-19;
 - (2) exposing the layer imagewise to actinic radiation to form an imaged cross-section, wherein the radiation is of sufficient intensity to cause substantial curing of the layer in the exposed areas;
 - 10 (3) coating a layer of the composition onto the previously exposed imaged cross-section;
 - (4) exposing said thin layer from step (3) imagewise to actinic radiation to form an additional imaged cross-section, wherein the radiation is of sufficient intensity to cause substantial curing and coloring of the thin layer in the exposed areas and to cause adhesion to the previously exposed imaged cross-section;
 - 15 (5) repeating steps (3) and (4) a sufficient number of times in order to build up the three-dimensional article.
21. The process according to claim 20, wherein the colored part is separated from the resin by cleaning.
- 20 22. The process according to claim 20 wherein the exposure energy used in step (4) is in the range of 10-250 mJ/cm².